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3-1 Calculate the specific volume of an air-vapor mixture in cubic meters per kilogram of dry air when the following conditions prevail:  $t = 30\text{ C}$ ,  $W = 0.015\text{ kg/kg}$ , and  $p = 90\text{ kPa}$ . Solution: Equation 3-4.  $t = 30\text{ C} + 273 = 303\text{ K}$   $R = 287\text{ J/kg.K}$   $p = 90\text{ kPa} = 90,000\text{ Pa}$  Equation 3-2  $t = 30\text{ C}$   $W = 0.015$   $p = 90\text{ kPa}$   $v = 1.35 - 0.15p$

## **or CLICK HERE**

Solution:  $q = mc_p(t_2 - t_1)$   $m = 2.5\text{ kg/s}$   $c_p = 1.0\text{ kJ/kg.K}$   $t_2 = 30\text{ C}$   $t_1 = -10\text{ C}$  CHAPTER 2 -

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THERMAL PRINCIPLES Page 2 of 5 Then,  $q = (2.5) (1.0) (30 + 10) q = 100 \text{ kw} - - - \text{Ans. 2-3.}$

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