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Water And Aqueous Systems Study Guide Water And Aqueous Systems Study SECTION 15.1 WATER AND ITS PROPERTIES (pages 445–449) 160 Guided Reading and Study Workbook Name _____ Date _____ Class _____ CHAPTER 15, Water and Aqueous Systems(continued) 6 Circle the letter next to each sentence

[PDF] Water And Aqueous Systems Study Guide

Chapter 15 (Water and Aqueous Systems) Test Study Guide The bonds between the hydrogen and oxygen atoms in a water molecule are polar covalent bonds. The covalent bonds are polar because the oxygen atom has a greater electronegativity than the hydrogen atoms. The bonds between adjacent water molecules are called hydrogen bonds.

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SECTION 15.1 WATER AND ITS PROPERTIES (pages 445–449)

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